

# Steam Turbine Generator

# Example of Power Generation Steam Turbine



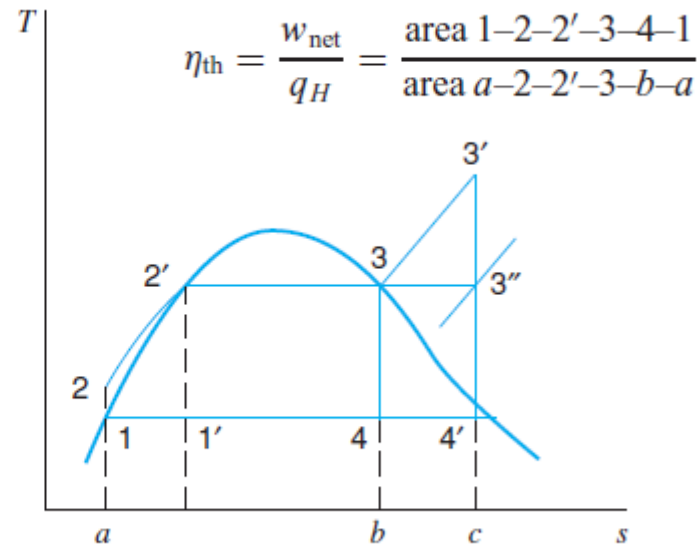
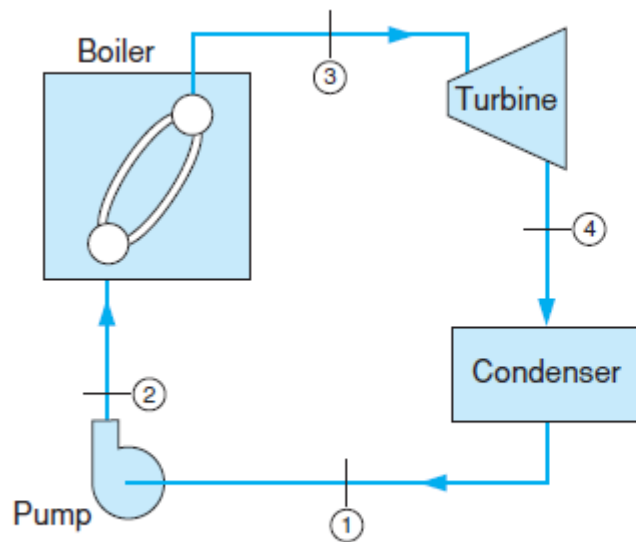
# Rankine Cycle

1–2: Reversible adiabatic pumping process in the pump

2–3: Constant-pressure transfer of heat in the boiler

3–4: Reversible adiabatic expansion in the turbine (or another prime mover such as a steam engine)

4–1: Constant-pressure transfer of heat in the condenser



- Adiabatic Process- Process that does not leave or take any heat
- No Heat loss/gain
- Thermal Eff= Net Work/Heat Input

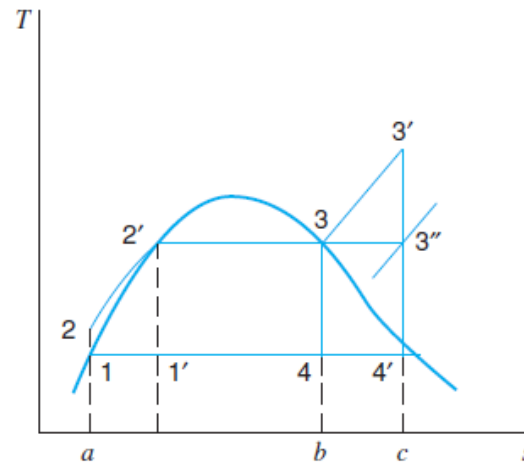
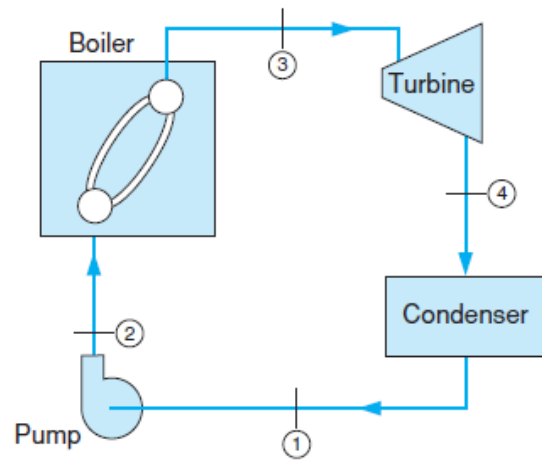
What is the difference between Pump and compressor?

Pump-- > for incompressible Liquid

Compressor → for Gas, air, compressible fluid

# Example Problem

Determine the efficiency of a Rankine cycle using steam as the working fluid in which the condenser pressure is 10 kPa. The boiler pressure is 2 MPa. The steam leaves the boiler as saturated vapor.



Given:

$P_3 = 2 \text{ MPa}$

$P_2$  is known = 2 MPa

$P_1 = 10 \text{ kPa}$

Steam leaves boiler at point 3  
Determine thermal Eff.

Pump work,  $W_p = h_2 - h_1$ ,  $h_2 = h_1 + w_p$

$$W_p = v(P_2 - P_1) = \text{Sp. Volume} \cdot (P_2 - P_1) = 0.001 \cdot (2000 - 10) = 2 \text{ kJ/kg}$$

Go to steam table and find

enthalpy of saturated water at  $P_1$  and  $P_2$ ,

$$h_1 = 191.75 \text{ kJ/kg}$$

$$h_2 = h_1 + W_p \text{ kJ/kg} = 191.75 + 2 = 193.8 \text{ kJ/kg}$$

# Boiler and Turbine Work

Here we will determine boiler heat input and turbine work  
And thermal efficiency

Now consider the boiler:

*Control volume:* Boiler.

*Inlet state:*  $P_2, h_2$  known; state fixed.

*Exit state:*  $P_3$  known, saturated vapor; state fixed.

**Analysis**

$$\text{Energy Eq.: } q_H = h_3 - h_2$$

**Solution**

Substituting, we obtain

$$q_H = h_3 - h_2 = 2799.5 - 193.8 = 2605.7 \text{ kJ/kg}$$

Net work= Turbine work-Pump work=792-2=790 kJ/kg

q<sub>H</sub>=Boiler heat input=2605kJ/kg

Efficiency= 790/2605=30.3%

Turning to the turbine next, we have:

*Control volume:* Turbine.

*Inlet state:* State 3 known (above).

*Exit state:*  $P_4$  known.

**Analysis**

$$\text{Energy Eq.: } w_t = h_3 - h_4$$

$$\text{Entropy Eq.: } s_3 = s_4$$

**Solution**

We can determine the quality at state 4 as follows:

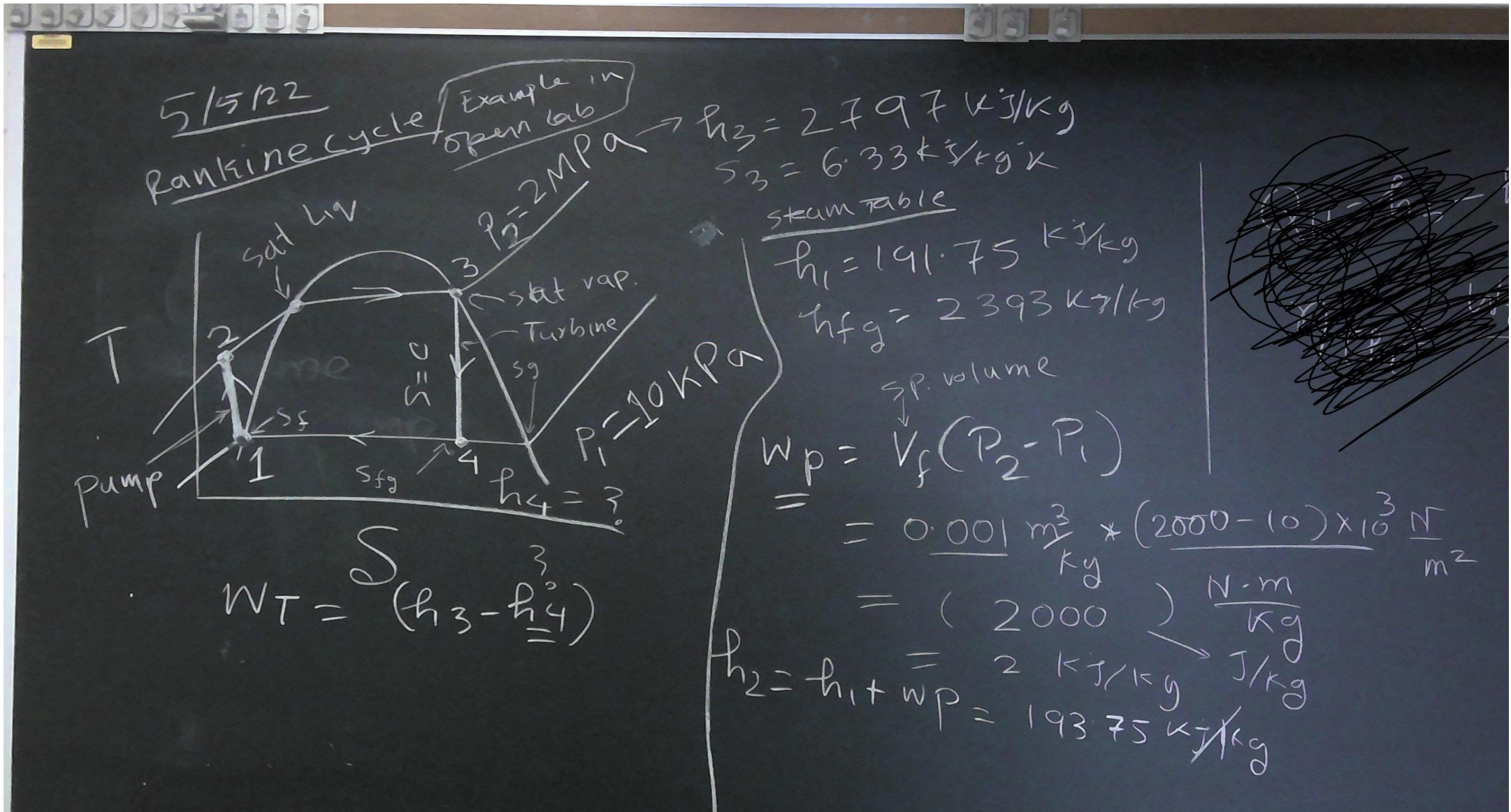
$$s_3 = s_4 = 6.3409 = 0.6493 + x_4 7.5009, \quad x_4 = 0.7588$$

$$h_4 = 191.8 + 0.7588(2392.8) = 2007.5 \text{ kJ/kg}$$

$$w_t = 2799.5 - 2007.5 = 792.0 \text{ kJ/kg}$$



# Solution to Rankine Cycle





$$Q_H = h_3 - h_2 = (2797 - 193.75) = \underline{\underline{2603 \text{ kJ/kg}}}$$

$$\eta_{Th} = \frac{W_T - W_P}{Q_H}$$

$$s = s_f + x * s_{fg}$$

$$h_4 \rightarrow s_2 = s_4 =$$

Quality of steam,  $x$

$$h = h_f + x * h_{fg}$$

$$h_4 = h_1 + x_4 * h_{fg}$$

$$s_4 = s_1 + x_4 * s_{fg} = s_3 = 6.33$$

$$s_4 = 6.33 = s_1 + x_4 * s_{fg} = 0.65 + x_4 * 7.5$$

$$\Rightarrow x_4 = 0.76 \text{ or } 76\%$$

$$h_4 = h_1 + x_4 * h_{fg} = 191.75 + 0.76 * 2393 = 2010 \text{ kJ/kg}$$

at state 1

$$s_f = 0.65$$

$$s_g = 8.15$$

$$s_{fg} = 7.5$$

$$W_T = 2797 - 2010 = 787 \text{ kJ/kg}$$

$$\eta_{Th} = \frac{787 - 2}{2603} = 0.3 \approx 30\%$$

# How to calculate Turbine Generator power?

Turbine work=  $\Delta h = h_2 - h_1 = \text{kJ/kg}$

How about kw? Turbine power=  $m(h_2 - h_1) =$

Partial Rankine Cycle Problem →

For example, steam mass flow rate,  $m = 10 \text{ kg/s}$ , from steam table  $h_2 = 3445 \text{ kJ/kg}$ ,  $h_1 = 2115 \text{ kJ/kg}$

Then Turbine power =  $10 \text{ kg/s} * (3445 - 2115) = 13300 \text{ kJ/s} = 13300 \text{ kW} = 13.3 \text{ MW}$

For example, a power generator has a 13.3 MW turbine, how much electricity you can produce if generator efficiency is 80%?

Electricity generated =  $13.3 * 0.8 = 10.6 \text{ MW}$ .