

More applications of exp and log

Lesson #16

MAT 1375 Precalculus

New York City College of Technology CUNY



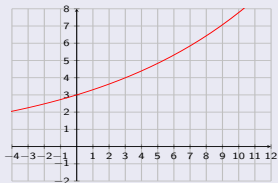
Exponential models

Exponential growth



Exponential models

Exponential growth

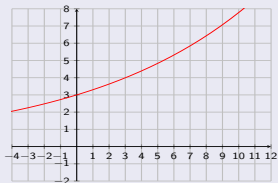


Examples:

-Population growth

Exponential models

Exponential growth



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- Population growth
- Investment at a fixed rate

Exponential models

Exponential growth



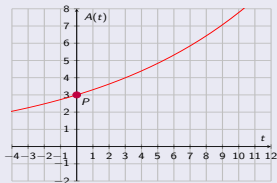
Examples:

- Population growth
- Investment at a fixed rate

Model: $A(t) = P \cdot e^{r \cdot t}$ or $A(t) = P \cdot (1 + r)^t$

Exponential models

Exponential growth



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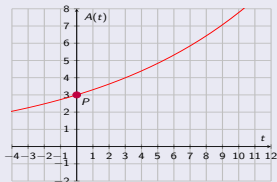
t = time

P = initial amount

$A(t)$ = amount at time t

Exponential models

Exponential growth



Examples:

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- Investment at a fixed rate

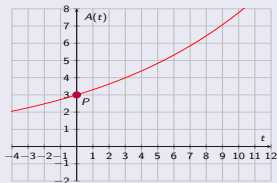
Model: $A(t) = P \cdot e^{r \cdot t}$ or $A(t) = P \cdot (1 + r)^t$
 $t =$ time $r =$ rate of growth

$P =$ initial amount

$A(t) =$ amount at time t

Exponential models

Exponential growth



Exponential decline



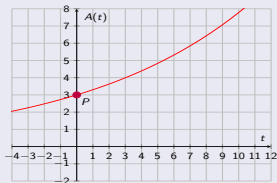
Examples:

- Population growth
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Model: $A(t) = P \cdot e^{r \cdot t}$ or $A(t) = P \cdot (1 + r)^t$
 t = time
 P = initial amount
 $A(t)$ = amount at time t
 $r > 0$: exponential growth
 $r < 0$: exponential decline

Exponential models

Exponential growth



Exponential decline



Examples:

- Population growth
- Investment at a fixed rate

Model: $A(t) = P \cdot e^{r \cdot t}$ or

t = time

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Examples:

- Population decline

$A(t) = P \cdot (1 + r)^t$

r = rate of growth or decay

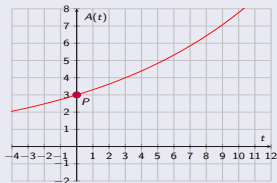
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Exponential models in applications

Exponential models

Exponential growth



Exponential decline



Examples:

- Population growth
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Model: $A(t) = P \cdot e^{r \cdot t}$ or

t = time

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Examples:

- Population decline
- Radioactive decay

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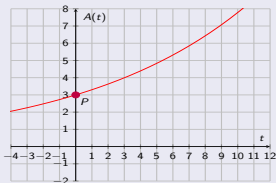
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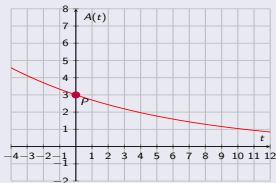
Exponential models in applications

Exponential models

Exponential growth



Exponential decline



Examples:

- Population growth
- Investment at a fixed rate

Model: $A(t) = P \cdot e^{r \cdot t}$ or

t = time

P = initial amount

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Examples:

- Population decline
- Radioactive decay

$A(t) = P \cdot (1 + r)^t$

r = rate of growth or decay

$r > 0$: exponential growth

$r < 0$: exponential decline

Note:

Both

$$A(t) = P \cdot e^{r \cdot t}$$

and

$$A(t) = P \cdot (1 + r)^t$$

could be used to model a situation.

For small r , these are close:

$$e^r \approx 1 + r$$

Use the model as stated.

If no particular model is stated use $A = P \cdot e^{r \cdot t}$.

Compound interest on an investment

Compound interest

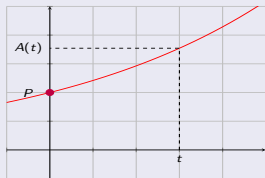
P = principal = (initial investment)

r = (continuous) interest rate

t = time

The current value of the investment, using a *continuous compounding*, is given by:

$$A(t) = P \cdot e^{r \cdot t}$$



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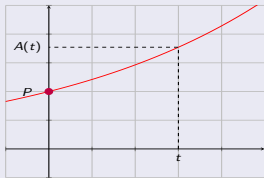
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- 1 A \$6,000 investment has an annual interest rate of 3% with a continuous compounding.
 - (a) How much is the investment worth after 7 years?
 - (b) How long will it take for the investment to grow to \$10,000?

Compound interest on an investment

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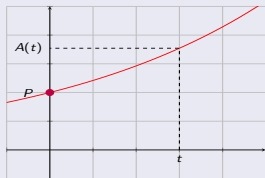
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- How much is the investment worth after 7 years?
- How long will it take for the investment to grow to \$10,000?

Solution:

(a) $P = 6,000$ (principal), $r = 3\% = 0.03$

$$\Rightarrow A(t) = 6000 \cdot e^{0.03 \cdot t}$$

$$t = 7 \Rightarrow A(7) = 6000 \cdot e^{0.03 \cdot 7} \approx 7402$$

It will be worth \$7,402.

Compound interest on an investment

Compound interest

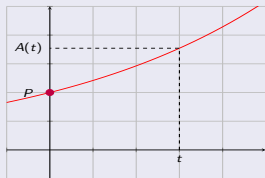
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(b) $A(t) = 10,000$

$$\Rightarrow 10,000 = 6,000 \cdot e^{0.03 \cdot t}$$

$$\Rightarrow \frac{10,000}{6,000} = e^{0.03 \cdot t}$$

$$\Rightarrow \ln\left(\frac{10,000}{6,000}\right) = 0.03 \cdot t$$

$$\Rightarrow t = \frac{\ln\left(\frac{10,000}{6,000}\right)}{0.03} \approx 17.0$$

It will take approximately 17 years.

Compound interest - exercises

- 2 A \$10,000 investment has an annual interest rate of 4.2% with a continuous compounding.
- (a) How much is the investment worth after 5 years?
- (b) How long will it take for the investment to double?

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- (b) How long will it take for the investment to double?

Solution:

(a) $P = 10,000$, $r = 4.2\% = 0.042$

$$\Rightarrow A(t) = 10000 \cdot e^{0.042 \cdot t}$$

$$t = 5$$

$$\Rightarrow A(5) = 10000 \cdot e^{0.042 \cdot 5} \approx 12,336.78$$

It will be worth \$12,336.78.

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$$\Rightarrow \frac{20,000}{10,000} = e^{0.042 \cdot t} \Rightarrow \ln(2) = 0.042 \cdot t$$

$$\Rightarrow t = \frac{\ln(2)}{0.042} \approx 16.5$$

It will take approximately 16.5 years for the investment to double its value.

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Compound n times per year

$$A(t) = P \cdot \left(1 + \frac{r}{n}\right)^{n \cdot t}$$

annual compounding: $n = 1$

semi-annual compounding: $n = 2$

quarterly compounding: $n = 4$

monthly compounding: $n = 12$

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- 3 \$5,000 is being invested at a rate of 2.7% compounded monthly. The value of the investment after t years is

$$A(t) = 5000 \cdot \left(1 + \frac{0.027}{12}\right)^{12 \cdot t}$$

What will the investment be worth after 15 years?

Compound interest - exercises

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Solution:

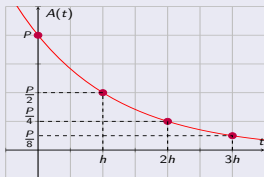
$$A(15) = 5000 \cdot \left(1 + \frac{0.027}{12}\right)^{12 \cdot 15} \approx 7,493.10$$

The investment will be worth \$7,493.10.

Radioactive decay

Radioactive decay

The *half-life* is the time it takes for a substance to decay to half of its original amount.



$$A(t) = P \cdot e^{r \cdot t}$$

For half-life h :

$$A(h) = \frac{P}{2}$$

$$A(2h) = \frac{P}{4}$$

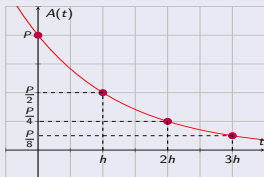
$$A(3h) = \frac{P}{8}$$

⋮

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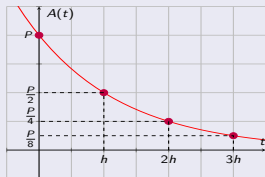
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- 1 A radioactive element has a half-life of 23 minutes. How long will it take for 85g of this element to have decayed to 5g?

Radioactive decay

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Solution:

$$P = 85\text{g}, r = ?$$

$$\Rightarrow A(t) = 85 \cdot e^{r \cdot t}$$

First, find r .

$$\text{Note: } A(23) = \frac{1}{2} \cdot 85 = 42.5$$

$$\Rightarrow 42.5 = A(23) = 85 \cdot e^{r \cdot 23}$$

$$\Rightarrow \frac{42.5}{85} = e^{r \cdot 23}$$

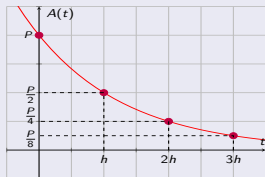
$$\Rightarrow \ln\left(\frac{1}{2}\right) = r \cdot 23$$

$$\Rightarrow r = \frac{\ln\left(\frac{1}{2}\right)}{23} \approx -0.03$$

Radioactive decay

Radioactive decay

The *half-life* is the time it takes for a substance to decay to half of its original amount.



$$A(t) = P \cdot e^{r \cdot t}$$

For half-life h :

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$$\Rightarrow \frac{42.5}{85} = e^{r \cdot 23}$$

$$\Rightarrow \ln\left(\frac{1}{2}\right) = r \cdot 23$$

$$\Rightarrow r = \frac{\ln\left(\frac{1}{2}\right)}{23} \approx -0.03$$

Then, for $A(t) = 5$:

$$\Rightarrow 5 = 85 \cdot e^{-0.03 \cdot t}$$

$$\Rightarrow \frac{5}{85} = e^{-0.03 \cdot t}$$

$$\Rightarrow \ln\left(\frac{5}{85}\right) = -0.03 \cdot t$$

$$\Rightarrow t = \frac{\ln\left(\frac{5}{85}\right)}{-0.03} \approx 94.4$$

It takes about 94.4 minutes.

Radioactive decay - exercises

- 2 Sodium-24 has a half-life of 14.96 hours. How long does it take for 30mg of Sodium-24 to decay to less than 2mg ?
- 3 How much is left after 26 weeks of 4.7lb of a radioactive substance, if the half-life of the substance is 8.9 weeks?

Radioactive decay - exercises

- 2 Sodium-24 has a half-life of 14.96 hours. How long does it take for 30mg of Sodium-24 to decay to less than 2mg?

Solution:

$$P = 30g, r = ?$$

$$\Rightarrow A(t) = 30 \cdot e^{r \cdot t}$$

First, find r .

$$\text{Note: } A(14.96) = \frac{1}{2} \cdot 30 = 15$$

$$\Rightarrow 15 = A(14.96) = 30 \cdot e^{r \cdot 14.96}$$

$$\Rightarrow 0.5 = e^{r \cdot 14.96}$$

$$\Rightarrow \ln(0.5) = r \cdot 14.96$$

$$\Rightarrow r = \frac{\ln(0.5)}{14.96} \approx -0.04633$$

Then, for $A(t) = 2$:

$$\Rightarrow 2 = 30 \cdot e^{-0.04633 \cdot t}$$

$$\Rightarrow \frac{2}{30} = e^{-0.04633 \cdot t}$$

$$\Rightarrow \ln\left(\frac{2}{30}\right) = -0.04633 \cdot t$$

$$\Rightarrow t = \frac{\ln\left(\frac{2}{30}\right)}{-0.04633} \approx 58.45$$

It takes about 58.45 hours.

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$$\Rightarrow A(t) = 30 \cdot e^{r \cdot t}$$

First, find r .

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$$\Rightarrow 15 = A(14.96) = 30 \cdot e^{r \cdot 14.96}$$

$$\Rightarrow 0.5 = e^{r \cdot 14.96}$$

$$\Rightarrow \ln(0.5) = r \cdot 14.96$$

$$\Rightarrow r = \frac{\ln(0.5)}{14.96} \approx -0.04633$$

Then, for $A(t) = 2$:

$$\Rightarrow 2 = 30 \cdot e^{-0.04633 \cdot t}$$

$$\Rightarrow \frac{2}{30} = e^{-0.04633 \cdot t}$$

$$\Rightarrow \ln\left(\frac{2}{30}\right) = -0.04633 \cdot t$$

$$\Rightarrow t = \frac{\ln\left(\frac{2}{30}\right)}{-0.04633} \approx 58.45$$

It takes about 58.45 hours.

- 3 How much is left after 26 weeks of 4.7lb of a radioactive substance, if the half-life of the substance is 8.9 weeks?

Solution:

$$P = 4.7, r = ?$$

$$\Rightarrow A(t) = 4.7 \cdot e^{r \cdot t}$$

First, find r .

$$\text{Note: } A(8.9) = \frac{1}{2} \cdot 4.7 = 2.35$$

$$\Rightarrow 2.35 = A(8.9) = 4.7 \cdot e^{r \cdot 8.9}$$

$$\Rightarrow 0.5 = e^{r \cdot 8.9}$$

$$\Rightarrow \ln(0.5) = r \cdot 8.9$$

$$\Rightarrow r = \frac{\ln(0.5)}{8.9} \approx -0.07788$$

$$\Rightarrow A(t) = 4.7 \cdot e^{-0.07788 \cdot t}$$

Then, for $t = 26$:

$$\Rightarrow A(26) = 4.7 \cdot e^{-0.07788 \cdot 26} \approx 0.62$$

After 26 weeks, only 0.62lb of the substance will be left.

